

HEAT and MASS TRANSFER

Introductory Concepts and Definitions

Ravikumar S

Department of Mechanical Engineering

ATMECE, Mysore

What is Heat Transfer?

“Energy in transit due to temperature difference.”

Thermodynamics tells us:

- How much heat is transferred (δQ)
- How much work is done (δW)
- Final state of the system

Heat transfer tells us:

- How (with what modes) δQ is transferred
- At what rate δQ is transferred
- Temperature distribution inside the body



MODES:

✓ Conduction

- needs matter
- molecular phenomenon (diffusion process)
- without bulk motion of matter

✓ Convection

- heat carried away by bulk motion of fluid
- needs fluid matter

✓ Radiation

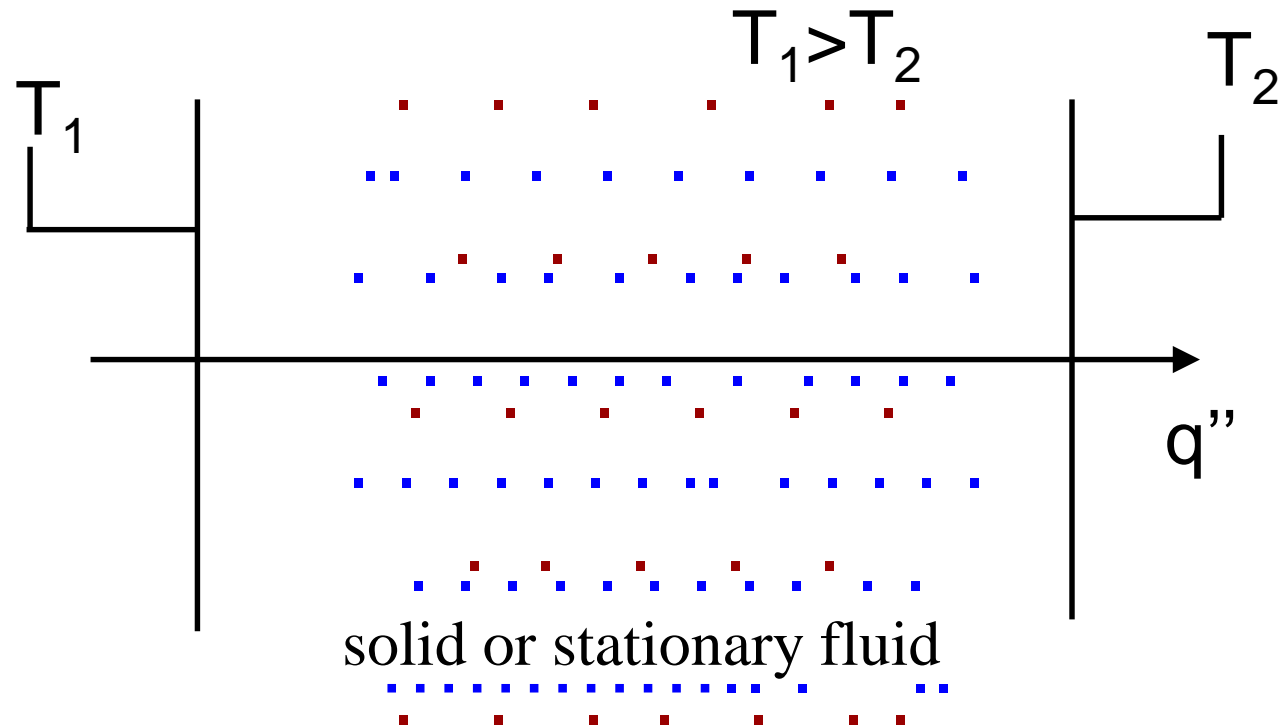
- does not needs matter
- transmission of energy by electromagnetic waves

APPLICATIONS OF HEAT TRANSFER

- ✓ Energy production and conversion
 - steam power plant, solar energy conversion etc.
- ✓ Refrigeration and air-conditioning
- ✓ Domestic applications
 - ovens, stoves, toaster
- ✓ Cooling of electronic equipment
- ✓ Manufacturing / materials processing
 - welding, casting, soldering, laser machining
- ✓ Automobiles / aircraft design
- ✓ Nature (weather, climate etc)

CONDUCTION

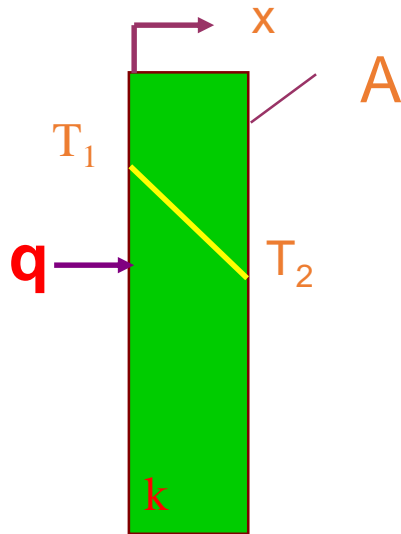
(Needs medium, Temperature gradient)



RATE:

q (W) or (J/s) (heat flow per unit time)

Conduction (contd...)



Rate equations (1D conduction):

□ Differential Form

$$q = -k A \frac{dT}{dx}, W$$

k = Thermal Conductivity, W/mK

A = Cross-sectional Area, m^2

T = Temperature, K or $^{\circ}C$

x = Heat flow path, m

□ Difference Form

$$q = k A (T_1 - T_2) / (x_1 - x_2)$$

Heat flux: $q'' = q / A = -k dT/dx$ (W/ m^2)

(negative sign denotes heat transfer in the direction of decreasing temperature)

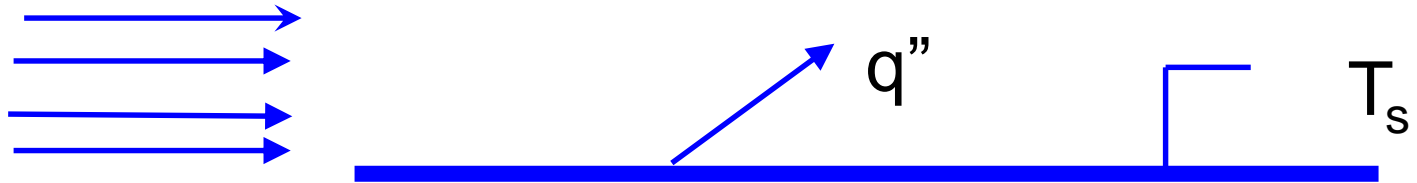
Conduction (contd...)

□ Example:

The wall of an industrial furnace is constructed from 0.2 m thick fireclay brick having a thermal conductivity of 2.0 W/mK. Measurements made during steady state operation reveal temperatures of 1500 and 1250 K at the inner and outer surfaces, respectively. What is the rate of heat loss through a wall which is 0.5 m by 4 m on a side ?

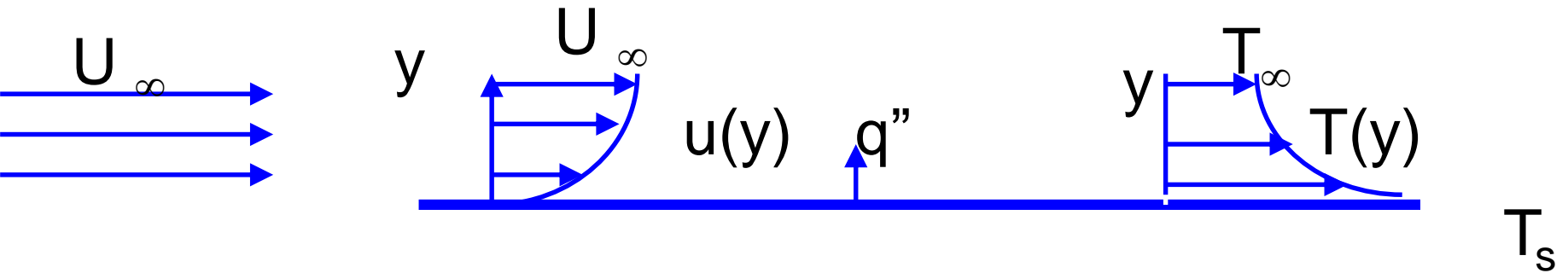
CONVECTION

moving fluid



❖ Energy transferred by diffusion + bulk motion of fluid

Rate equation (convection)



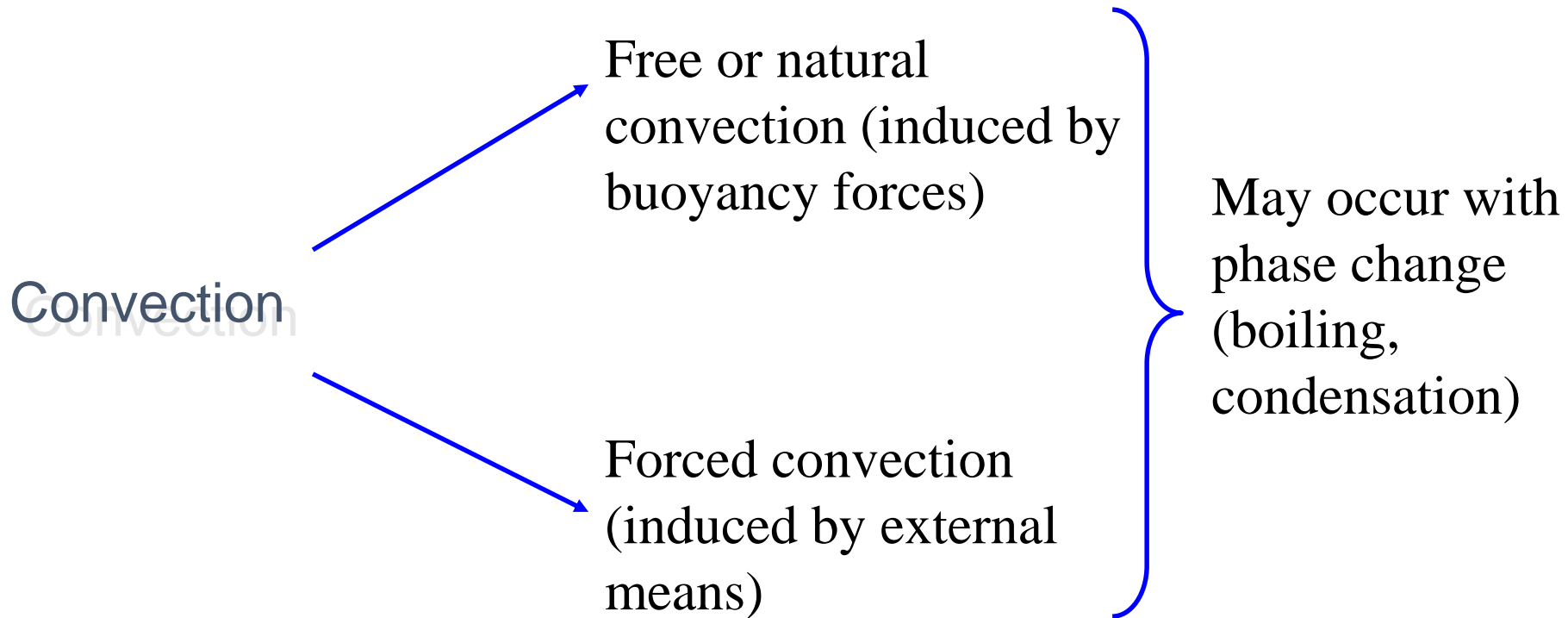
$$\text{Heat transfer rate } q = hA(T_s - T_{\infty}) \quad \text{W}$$

$$\text{Heat flux } q'' = h(T_s - T_{\infty}) \quad \text{W} / \text{m}^2$$

h =heat transfer co-efficient ($\text{W} / \text{m}^2\text{K}$)

(not a property) depends on geometry ,nature of flow,
thermodynamics properties etc.

Convection (contd...)



Convection (contd...)

Typical values of h ($\text{W/m}^2\text{K}$)

Free convection

gases: 2 - 25

liquid: 50 - 100

Forced convection

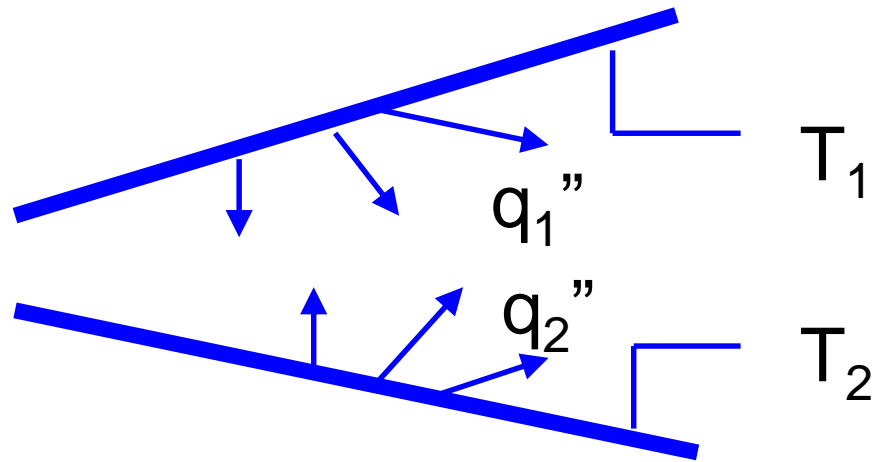
gases: 25 - 250

liquid: 50 - 20,000

Boiling/Condensation

2500 - 100,000

RADIATION



RATE:

$q(\text{W})$ or (J/s) Heat flow per unit time.

Flux : $q'' (\text{W/m}^2)$

Rate equations (Radiation)

RADIATION:

Heat Transfer by electro-magnetic waves or photons(no medium required.)

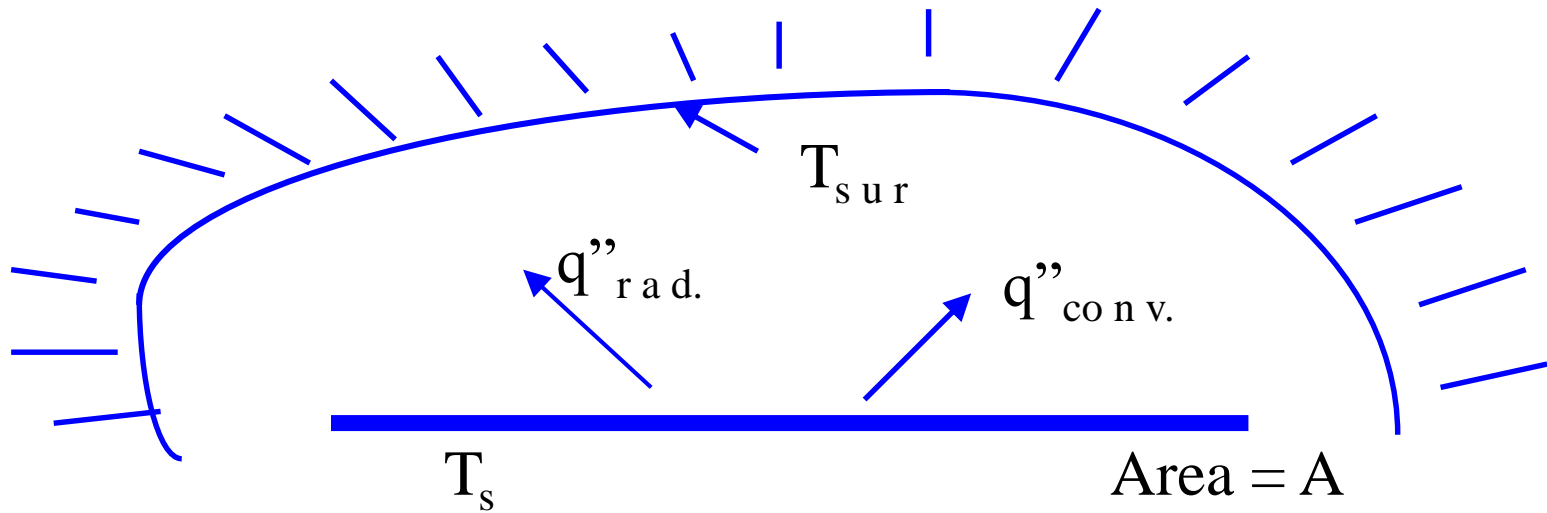
Emissive power of a surface (energy released per unit area):

$$E = \epsilon \sigma T_s^4 \text{ (W/ m}^2\text{)}$$

ϵ = emissivity (property).....

σ = Stefan-Boltzmann constant

Rate equations (Contd....)



Radiation exchange between a large surface and
surrounding

$$Q''_{rad} = \epsilon \sigma (T_s^4 - T_{sur}^4) \text{ W/ m}^2$$

Radiation (contd...)

□Example:

An uninsulated steam pipe passes through a room in which the air and walls are at 25°C . The outside diameter of pipe is 80 mm, and its surface temperature and emissivity are 180°C and 0.85, respectively. If the free convection coefficient from the surface to the air is $6 \text{ W/m}^2\text{K}$, what is the rate of heat loss from the surface per unit length of pipe ?